

## AN EVALUATION OF THE REDOX CYCLING POTENCIES OF PARAQUAT AND NITROFURANTOIN IN MICROSOMAL AND LUNG SLICE SYSTEMS

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**Abstract**—The redox cycling abilities of the pulmonary toxins paraquat and nitrofurantoin have been compared with those of the potent redox cyclers, diquat and menadione in lung and liver microsomes by using the oxidation of NADPH and consumption of oxygen. The relative potencies of these compounds to undergo redox cycling were in the order: diquat  $\approx$  menadione  $\gg$  paraquat  $\approx$  nitrofurantoin. This was partly attributed to the much lower affinity ( $K_m$ ) of lung and liver microsomes for paraquat and nitrofurantoin than for diquat and menadione. The potential to redox cycle was assessed in an intact cellular system by determining the oxygen consumption of rat lung slices in the presence ( $10^{-6}$ ,  $10^{-5}$  and  $10^{-4}$  M) or absence of each of the four substrates. At concentrations of paraquat ( $10^{-5}$  M) known to be accumulated by lung slices, a small but significant stimulation of lung slice oxygen uptake was observed. Nitrofurantoin ( $10^{-4}$ – $10^{-6}$  M) did not affect lung slice oxygen uptake in lung slices, an observation consistent with its being a poor redox cycling compound, which is not actively accumulated into lung cells. This data has important implications in assessing the risk of exposure to paraquat. Low levels of paraquat would not be expected to cause lung damage because insufficient compound is present in the lung to exert its toxicity by redox cycling (due to the high  $K_m$  observed).

Cellular damage resulting from the injurious effects of free radicals may lead to a number of conditions including cancer, ageing and drug-induced toxicity [1–3]. It has been proposed that the pulmonary toxicities of the herbicide, paraquat [2, 4] and the antibiotic, nitrofurantoin [2, 5] are related to their ability to form free radicals which can react with molecular  $O_2$  to form the superoxide anion radical ( $O_2^-$ ). Superoxide anion radicals or other activated oxygen radical species formed from them, such as hydroxyl radicals ( $OH^\bullet$ ) [6, 7] and singlet  $O_2$  ( $^1O_2$ ) [8] may damage cells by a number of mechanisms including peroxidation of polyunsaturated fatty acids [9, 10] or depletion of cellular reducing equivalents such as NADPH [11]. Furthermore, there may be a disturbance of cellular thiol status, either by depletion of intracellular reduced glutathione [12] or formation of protein-mixed disulphides. Protection against “oxidative stress” is normally afforded by cellular antioxidant defences such as superoxide dismutase, catalase, the glutathione peroxidase–glutathione reductase enzyme systems [4, 13], vitamin E [14] and vitamin C [15]. Cellular injury results when pro-oxidant effects exceed the antioxidant defence capability.

Paraquat (1,1'-dimethyl-4,4'-bipyridinium) has gained notoriety largely as a consequence of self-inflicted poisoning [16–18]. Lung damage is a prominent pathological feature and is often the primary cause of death of the patient [19]. Much work has been carried out to elucidate the mechanism of toxicity of paraquat in order to develop a specific antidote. However, the precise mode of toxic action is still contentious. Some details of the mechanism of toxicity have been clearly established. Paraquat is

selectively accumulated into the lung compared with other tissues by an energy-dependent transport system which is probably present for the accumulation of endogenous polyamines [20, 21] and/or cystamine [22]. This accumulation occurs primarily in alveolar type I and II cells and in Clara cells [23] and these cell types are selectively damaged by paraquat in comparison with other lung cell types [23]. It is generally accepted that the critical biochemical event in paraquat toxicity is the cyclical one electron reduction and reoxidation of paraquat utilizing cellular reducing equivalents and producing superoxide anion radical [2, 4].

Patients treated for prolonged periods with nitrofurantoin (*N*-(5-nitro-2-furfurylidene)-1-aminohydantoin) have been reported to develop fatal lung fibrosis similar to that induced by paraquat [24, 25], which has led to the suggestion that both compounds produce lung damage through a similar mechanism [26]. Nitrofurantoin undergoes redox cycling under aerobic conditions with appropriate electron donors [27] but, unlike paraquat, it is not selectively accumulated into the lungs and it binds covalently to cellular macromolecules [28]. Although both compounds redox cycle in *in vitro* systems [26, 29], their redox cycling potencies have not been quantified. We have therefore determined the abilities of paraquat and nitrofurantoin to redox cycle in comparison to two other potent redox cycling compounds, diquat and menadione, in an attempt to relate redox cycling ability to toxicity.

Diquat (*N,N'*-ethylene-2,2'-bipyridinium), a herbicide structurally related to paraquat, produces minimal lung damage *in vivo* [30], probably due to its inability to be taken up by the pulmonary

polyamine uptake system and so accumulate in the lung [31]. Menadione [2-methyl-1,4-naphthoquinone] is a potent redox cyler with liver microsomes [32] and has been extensively utilized as an investigative tool in biological studies involving oxidative stress.

The role of redox cycling in mediating the toxicity of paraquat and nitrofurantoin has been widely investigated [4, 5]. Under appropriate experimental conditions, the free radical forms of both these compounds and activated  $O_2$  species produced following their autoxidation have been demonstrated [27]. Rat liver and lung microsomes have been shown to mediate the one-electron reduction of paraquat, nitrofurantoin, diquat and menadione probably via NADPH-cytochrome P-450 reductase [10, 26, 33], and were used to study the redox cycling potencies of these compounds. However the ease by which they undergo redox cycling is unclear. We have determined the relative potencies of paraquat and nitrofurantoin to redox cycle in microsomal systems by measuring their utilization of  $O_2$  and NADPH in comparison to two potent redox cycling compounds, diquat and menadione. Using liver or lung microsomes, the relative potencies of the compounds to undergo redox cycling were: diquat = menadione  $\gg$  paraquat = nitrofurantoin. With lung slices only small increases in  $O_2$  consumption were observed in the presence of any of these compounds indicating that such  $O_2$  utilization comprised only a small percentage of total lung  $O_2$  consumption.

#### MATERIALS AND METHODS

Nitrofurantoin was purchased from the Sigma Chemical Co. (St Louis, MO, U.S.A.). Menadione was obtained from Aldrich Chemical Co. (Gillingham, Dorset, U.K.). Paraquat and diquat were kindly provided by I.C.I. Plant Protection Division (Surrey, U.K.). Other chemicals were obtained either from the Sigma Chemical Co. or B.D.H. (Poole, Dorset, U.K.).

**Preparation of microsomes.** Washed hepatic and pulmonary microsomes were prepared from fed, male Wistar-derived Alderley Park rats, essentially by the method of Ernster *et al.* [34]. The organs were perfused with ice-cold 1.15% KCl, minced and homogenized with a Polytron homogeniser at full speed for 30 and 45 sec for liver and lung, respectively. The washed microsomal pellet was resuspended in Tris-EGTA buffer (100 mM Tris, 5 mM EGTA; pH 7.4). Microsomal protein was determined by the method of Lowry *et al.* [35].

**NADPH oxidation.** The oxidation of NADPH at 37° was followed at 340 nm on a Shimadzu MPS 2000 spectrophotometer as previously described [12]. Liver and lung microsomes were used at concentrations of 0.2 and 0.4 mg protein/mL, respectively. Paraquat and diquat were dissolved in buffer (100 mM Tris, 5 mM EGTA; pH 7.4), whilst nitrofurantoin and menadione were dissolved in dimethylsulphoxide (1%) (DMSO).

The activity of microsomal NADPH-cytochrome *c* reductase was determined spectrophotometrically by following the reduction of cytochrome *c* (Sigma type III) by NADPH at 550 nm on a Shimadzu MPS

200 spectrophotometer. The specific activity of the enzyme was calculated using an extinction coefficient of 19.6/mM/cm [36].

**$O_2$  consumption.** The microsomal consumption of  $O_2$  was measured on a Yellow Spring Clark oxygen electrode maintained at 37°, as previously described [37]. Microsomes were used at a concentration of 0.2 mg protein/mL.

**Rat lung slices and respirometry studies.** Rat lung slices were prepared from fed, 180–200 g male, Wistar-derived, Alderley Park rats [38].  $O_2$  consumption of rat lung slices (35–45 mg wet wt) at 37° was measured using a Gilson respirometer [39]. Paraquat and diquat were dissolved in Krebs–Ringer phosphate buffer, pH 7.4, whilst nitrofurantoin and menadione were dissolved in 0.1% DMSO.

#### RESULTS

##### NADPH oxidation and $O_2$ consumption

The initial rates of rat liver and lung microsomal NADPH oxidation were stimulated by all four compounds. However, the stimulation produced by diquat and menadione was greater than that produced by paraquat and nitrofurantoin (Fig. 1a and b). Similarly, all four compounds stimulated the initial rates of rat liver microsomal  $O_2$  consumption, with diquat and menadione producing the greatest stimulation (Fig. 2).

To determine the stoichiometry of these reactions, the amounts of NADPH oxidized and  $O_2$  utilized by each of the four substrates were calculated after 1 and 5 min incubation. After 1 min, neither paraquat nor nitrofurantoin caused a greater than stoichiometric oxidation of NADPH (Fig. 1) and consumption of  $O_2$  (Fig. 2). In contrast, menadione and diquat (5–10  $\mu$ M), oxidized NADPH and utilized  $O_2$  in amounts that were approximately 10-fold their concentration in the systems (Figs 1 and 2). However, after 5 min of incubation, both paraquat and nitrofurantoin (5–10  $\mu$ M) utilized  $O_2$  and oxidized NADPH in amounts approximately 4–5-fold greater than the concentration of substrate present in the incubation (Table 1). Diquat and menadione again caused similar disproportionate oxidation of NADPH and consumption of  $O_2$  (Table 1).

##### Kinetic parameters

Lung and liver microsomal NADPH oxidation were studied in the presence of various concentrations of paraquat (20  $\mu$ M–5 mM), diquat (10  $\mu$ M–1 mM), nitrofurantoin (20  $\mu$ M–100  $\mu$ M) and menadione (2  $\mu$ M–60  $\mu$ M).  $K_m$  and  $V_{max}$  values were calculated from Lineweaver–Burk, Eadie–Hofstee, Hanes–Woelf and Direct Linear plots using a computer programme (ENZPACK) (Table 2). Values of  $K_m$  for paraquat and nitrofurantoin were of a similar order of magnitude and were about 20-fold greater than those for diquat and menadione.  $V_{max}$  values were either expressed per mg of microsomal protein or were standardized to a unit of specific activity of microsomal NADPH-cytochrome *c* reductase.  $V_{max}$  for nitrofurantoin in liver microsomes was lower than for the other compounds (Table 2).

##### $O_2$ consumption of lung slices

In order to obtain some indication of their ability

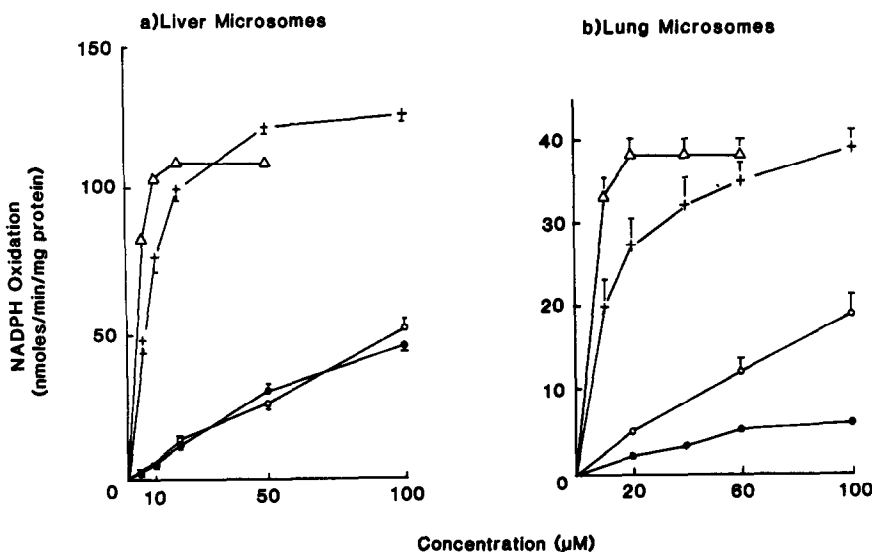


Fig. 1. Effect of substrates on initial rates of NADPH oxidation. Oxidation of NADPH by liver and lung microsomes incubated with paraquat (○—○), NF (●—●), DQ (+—+) and menadione (Δ—Δ) was followed at 340 nm at 37° for 1 min. Values shown have been corrected for basal rates of microsomal NADPH oxidation ( $12.17 \pm 0.33$  nmol NADPH/mg protein/min). Each point represents the mean  $\pm$  SE of three experiments with three determinations per substrate concentration per experiment, except for menadione in liver microsomes and nitrofurantoin in lung microsomes.

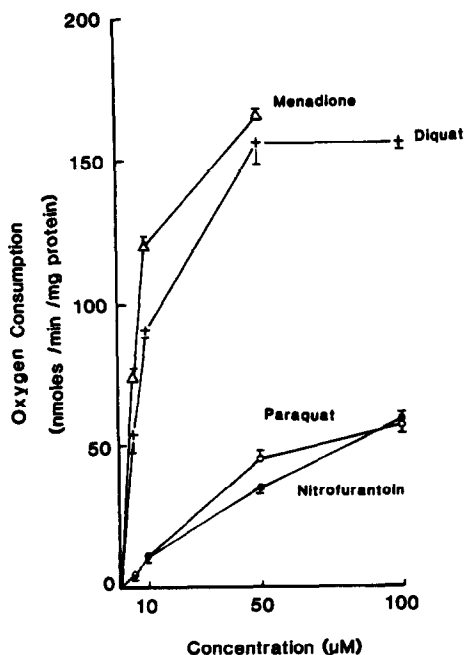


Fig. 2. Effect of substrates on initial rates of  $O_2$  consumption. The initial rates of liver microsomal  $O_2$  consumption incubated with paraquat (○—○), NF (●—●), DQ (+—+) and menadione (Δ—Δ) were determined at 37° for 1 min. Values shown have been corrected for basal rates of microsomal  $O_2$  consumption ( $15.95 \pm 0.53$  nmol  $O_2$ /mg protein/min). Each point represents the mean  $\pm$  SE of three separate experiments with two determinations per substrate concentration per experiment.

to redox cycle in a cellular system, the  $O_2$  consumption of lung slices in the presence or absence of paraquat, diquat, nitrofurantoin or menadione was followed for 4 hr of incubation at 37°. In control lung slices, the rate of  $O_2$  uptake was linear over this period (90–120  $\mu$ L  $O_2$  100/mg wet wt/hr).

A significant increase ( $P < 0.01$ ) in the  $O_2$  uptake of lung slices was observed with 10  $\mu$ M paraquat (Fig. 3a). Although 100  $\mu$ M paraquat also caused a stimulation in lung slice  $O_2$  uptake, the increase observed was smaller (Fig. 3a). No significant increase in lung slice  $O_2$  uptake was observed when slices were incubated with a lower concentration of paraquat (1  $\mu$ M) (Fig. 3a). Nitrofurantoin (1–100  $\mu$ M) did not produce any significant effects on lung slice  $O_2$  consumption (Fig. 3b).

Low concentrations of menadione and diquat (1  $\mu$ M and 10  $\mu$ M) did not affect the lung slice  $O_2$  uptake (Fig. 3c and d). However, 100  $\mu$ M diquat produced a small, but significant increase ( $P < 0.01$ ) in  $O_2$  consumption after 1 hr (Fig. 3d). In contrast, 100  $\mu$ M menadione produced a significant time-dependent depression ( $P < 0.01$ ) in the  $O_2$  uptake of lung slices when compared to controls (Fig. 3c).

## DISCUSSION

### Microsomal NADPH and $O_2$ utilization

Diquat and menadione elicited a nonstoichiometric increase in the utilization of NADPH and  $O_2$  in liver and lung microsomal systems (Figs 1 and 2). In contrast, although both paraquat and nitrofurantoin stimulated microsomal NADPH oxidation and  $O_2$  consumption, neither of the compounds caused a nonstoichiometric increase in the

Table 1. Amounts of NADPH and O<sub>2</sub> utilized after 5 min of substrate addition to liver microsomal systems

Substrate concentration (nmol)	nmol NADPH oxidized/ mg protein/5 min				nmol O <sub>2</sub> consumed/ mg protein/5 min			
	PQ	NF	DQ	Md	PQ	NF	DQ	Md
5	22.6	20.1	187.7	246.2	27.3	27.7	200.8	204.8
10	42.4	38.6	244.6	259.6	44.2	16.9	302.3	341.9
20	76.9	65.8	273.9	244.2	ND	ND	ND	ND
50	128.3	129.2	280.8	236.1	134.0	104.5	443.4	485.5
100	198.4	193.5	273.3	ND	199.2	184.9	449.0	ND

ND: not determined.  
Values shown have been corrected for the amount of NADPH oxidized in 5 min (53.9 ± 1.7 nmol NADPH/mg protein) and the amount of O<sub>2</sub> utilized in 5 min (81.6 ± 1.7 nmol O<sub>2</sub>/mg protein) by liver microsomes in the absence of substrates. Each value represents the means of three different experiments.

Table 2. Kinetic parameters of microsomal NADPH oxidation with various substrates

Substrates	Rat liver microsomes			Rat lung microsomes		
	K <sub>m</sub> (μM)	V <sub>max</sub> <sup>*</sup>	V <sub>max</sub> <sup>†</sup>	K <sub>m</sub> (μM)	V <sub>max</sub> <sup>*</sup>	V <sub>max</sub> <sup>†</sup>
PQ	143 ± 13.7	148 ± 3	0.73 ± 0.02	198 ± 6.2	54 ± 0.9	0.79 ± 0.01
DQ	6.2 ± 0.7	124 ± 2.0	0.60 ± 0.02	17 ± 0.4	45 ± 0.3	0.88 ± 0.01
Menadione	5.1 ± 0.2	132 ± 1.5	1.27 ± 0.33	6.9 ± 0.4	45 ± 0.9	0.69 ± 0.01
NF‡	102.6 ± 19.4	53.2 ± 6.6	0.28 ± 0.03	ND	ND	ND

<sup>\*</sup> V<sub>max</sub>: nmol NADPH oxidized per mg microsomal protein per min.  
<sup>†</sup> V<sub>max</sub>: nmol NADPH oxidized per unit specific activity of NADPH-cytochrome c reductase.  
<sup>‡</sup> NF: values were obtained from extrapolated results. ND: not determined.  
Results are means ± SE of four different methods of calculations (Lineweaver-Burk, Eadie-Hofstee, Hanes-Woolf and Direct Linear) on results from three different experiments per substrate used. The specific activity of NADPH-cytochrome c reductase for liver and lung microsomes was 198 ± 3.5 and 57 ± 5.8 nmol cytochrome c reduced/mg protein/min, respectively.

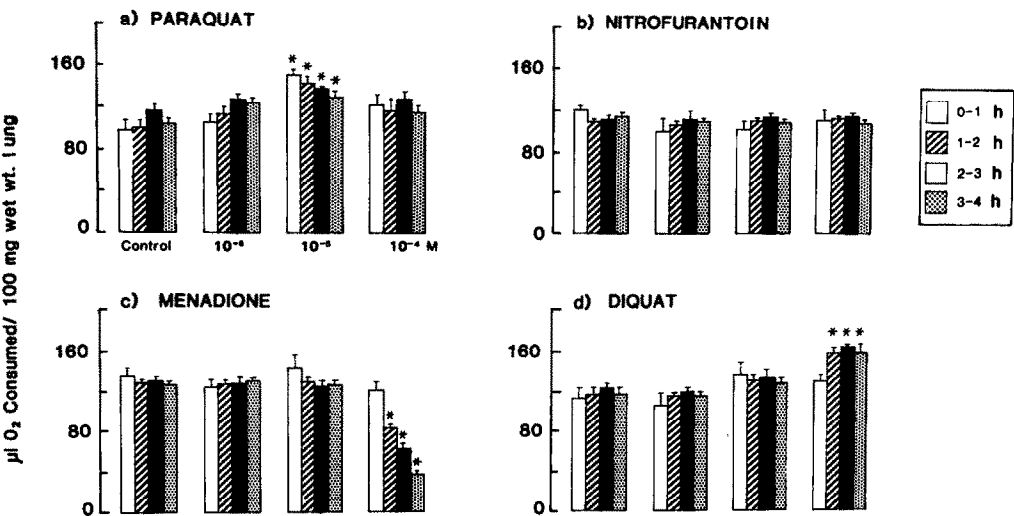


Fig. 3. O<sub>2</sub> consumption of rat lung slices. Lung slices were incubated with (a) paraquat, (b) nitrofurantoin, (c) menadione and (d) DQ at concentrations of 10<sup>-6</sup>, 10<sup>-5</sup> and 10<sup>-4</sup> M at 37°. Control slices were incubated either in buffer or in 0.1% DMSO. Values indicate the hourly increment in O<sub>2</sub> consumption and represent the means of three experiments performed on six rats. \*Significantly different from control (P < 0.01, t-test).

utilization of NADPH and  $O_2$  after 1 min of incubation (Figs 1 and 2). However, after 5 min of incubation, paraquat and nitrofurantoin utilized a disproportionate amount of  $O_2$  and NADPH when compared to the amounts of substrate present (Table 1). Thus it is apparent that in microsomal systems, the relative potency of the compounds to redox cycle was diquat  $\approx$  menadione  $\gg$  paraquat  $\approx$  nitrofurantoin.

The poor redox cycling of paraquat and nitrofurantoin was accurately reflected in the kinetic parameters (Table 2). Values of  $K_m$  for diquat and menadione were comparable and much lower than for paraquat (in lung and liver microsomes) and nitrofurantoin (in liver microsomes) (Table 2). The  $K_m$  value, which reflects the affinity of NADPH-cytochrome *c* reductase for the substrate is a composite of two reactions; one-electron reduction of the substrate, utilizing NADPH, and probably catalysed by NADPH-cytochrome P-450 reductase and autoxidation of the reduced substrate to yield the parent compound with concomitant production of activated  $O_2$  species. Since interaction of one-electron reduced forms of substrates with molecular  $O_2$  is diffusion limited [2, 26, 40, 41], we can assume that the  $K_m$  values reflect the one-electron reduction reactions only. Thus, NADPH-cytochrome *c* reductase appears to have much lower affinities for paraquat and nitrofurantoin than for diquat and menadione.  $V_{max}$  values for the different substrates were comparable in both lung and liver microsomes, except for nitrofurantoin, which was lower (Table 2). Values of  $V_{max}$  for paraquat, diquat and menadione were similar in both lung and liver microsomes when expressed on the basis of NADPH-cytochrome *c* reductase activity (Table 2) because the activity of the pulmonary NADPH-cytochrome P-450 reductase is lower than that of the liver [42].

#### Lung slice $O_2$ consumption

In order to assess the potential contribution of the redox cycling to the toxicity of these compounds *in vivo*, the effects of the compounds on  $O_2$  utilization was studied in lung slices, a model system which retained intact cellular structure. A significant increase in lung slice  $O_2$  uptake was observed with  $10^{-5}$  M paraquat but not with  $10^{-6}$  M paraquat (Fig. 3a). As paraquat has been shown in the present study to be a poor redox cyclers, high intracellular concentrations are necessary to produce a significant increase in lung slice  $O_2$  uptake as a result of redox cycling. These would be achieved by active accumulation into susceptible cell types. In fact, with  $10^{-6}$  M paraquat, rat lung slices have been shown to accumulate ten times the amount of paraquat intracellularly after 2 hr of incubation at 37° [43]. Thus, the stimulation in  $O_2$  uptake observed with  $10^{-5}$  M paraquat could be due to its redox cycling in susceptible cells. At higher concentrations, redox cycling may lead to toxicity, causing a decrease in  $O_2$  consumption of lung slices incubated with  $10^{-4}$  M compared to those incubated with  $10^{-5}$  M paraquat.

Lung slices incubated with  $10^{-6}$  M and  $10^{-5}$  M menadione and diquat did not significantly affect lung slice  $O_2$  uptake (Fig. 3c and d). However, lung slices incubated with  $10^{-4}$  M diquat utilized

significantly more  $O_2$  than the controls. At this concentration, enough redox active diquat molecules may have passively diffused into susceptible cells to redox cycle without exceeding cellular defences (Fig. 3d). In contrast, menadione at the same concentration caused significant inhibition of lung slice  $O_2$  uptake (Fig. 3c), probably due to toxicity to susceptible cells, either by redox cycling, and/or covalent binding to cellular macromolecules [12]. The inability of lower concentrations of menadione ( $10^{-6}$  and  $10^{-5}$  M) to stimulate lung slice  $O_2$  uptake may be due to toxicity of the quinone to lung cells, which may primarily involve a covalent binding, rather than a redox cycling mechanism.

Nitrofurantoin did not affect the lung slice  $O_2$  uptake (Fig. 3b), an observation consistent with its being a poor redox cyclers as well as it not being actively taken up into the lungs. However, other factors such as mitochondrial respiration, besides redox cycling may effect lung slice  $O_2$  consumption. Nitrofurantoin has been shown to inhibit liver mitochondrial respiration [44], as well as to bind covalently to cellular macromolecules and these observations may also partially explain its apparent lack of effect on lung slice  $O_2$  uptake.

#### Implications for risk assessment

The kinetic parameters that were established for the redox cycling of paraquat in the lung may have implications in defining the toxicity of paraquat and assessing the hazard that results when animals or man are exposed to small quantities of the herbicide. The data suggests that the intrinsic toxicity of paraquat is relatively low. This is particularly true when compared with the related dipyridyl, diquat, which, although not nearly as toxic to the lung, is quantitatively a much more effective redox cycling agent (compare  $K_m$ s for microsomal NADPH utilization between paraquat and diquat, Table 2). However, in lung slices paraquat causes a greater increase in  $O_2$  consumption than diquat indicating that with intact lung slices it is more able to redox cycle. This is consistent with previous findings and illustrates the absolute importance of the selective uptake of paraquat into specific lung cells. Without this process, the lung would not be able to accumulate toxic levels of paraquat. The absence of a similar accumulation process for diquat is the most probable explanation as to why this herbicide, which is the more potent redox cyclers, does not cause significant lung damage *in vivo*.

The  $K_m$  for NADPH oxidation by paraquat (198  $\mu$ M) may also provide an insight for the reason why paraquat does not provoke insidious pulmonary toxicity in individuals exposed to low levels of the herbicide. Until the intracellular concentration of paraquat reaches concentrations that significantly reduce NADPH levels (and concomitantly generate  $O_2^-$ ) no toxicity will be observed. The relatively high  $K_m$  found with paraquat for this process means that significant toxicity will only occur when the lung is able to accumulate these levels of paraquat from the plasma. This has been observed by others [45] since it can be calculated that unless a critical amount of paraquat is present in the plasma no toxicity will be produced. In other words, unless a threshold

concentration is reached in the lung, no damage results. This data provides some quantifiable evidence that this is the case since the damaging consequence of redox cycling will not be manifest until the intracellular concentration of paraquat is well into the micromolar range.

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